

# NUMERICAL STUDY ON NO<sub>x</sub> EMISSION REDUCTION MECHANISM OF HCCI MARINE DIESEL ENGINE FOR IMO TIER III EMISSION LIMITS

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## SUMMARY

With the advantages of ultra-low emissions of oxides of nitrogen (NO<sub>x</sub>) and high thermal efficiency, the homogeneous charge compression ignition (HCCI) mode applied to marine diesel engine is expected to be one of the technical solutions to meet the International Maritime Organization (IMO) MARPOL73/78 Convention-Annex VI Amendment Tier III requirement. According to the NO<sub>x</sub> chemical reaction mechanism, taking a marine diesel engine as the application object, the numerical study on the NO<sub>x</sub> formation characteristics of n-heptane for HCCI combustion process is performed. The results indicate that NO is usually the main component in the generation and emissions of NO<sub>x</sub> with the n-heptane HCCI mode. The combustor temperature plays more important role in the proportion of NO generation and emission. Compared with the experimental data of conventional marine diesel engine, the emission reduction rate of NO<sub>x</sub> can achieve an average of more than 95% in using HCCI technology.

## 1. INTRODUCTION

With the development of global economy and shipping industry, air pollution from marine diesel engine is becoming serious increasingly, especially for the air quality of coastal areas. These situations have attracted wide international attention. In the combustion process of diesel engine, the pollutant emissions mainly consist of oxides of sulphur (SO<sub>x</sub>), oxides of nitrogen (NO<sub>x</sub>), gaseous hydrocarbon (HC), carbon monoxide (CO) and particulate matters (PM). Among them, SO<sub>x</sub> and NO<sub>x</sub> are the components which are the biggest emitters, hazardous environmental and most tightly controlled. In recent years, it is estimated that the NO<sub>x</sub> emissions from shipping occupied 18% to 30% of the total worldwide emissions according to the related statistical information on this topic (Wei, Y., 2013, Eyring, V., 2005, Øyvind, E., 2003). To control and reduce the air pollution from marine diesel engine, the Marine Environmental Protection Committee (MEPC) of the International Maritime Organization (IMO) has agreed upon progressively stricter limitations for NO<sub>x</sub> emissions from vessels based on their date of engine installation with the standards of Tier I, Tier II and Tier III in October 2008. These standards have taken effect beginning on July 1, 2010. As shown in Table 1, the Tier III standard is the strictest mandatory requirement applied for the engine whose installation date later than January 1, 2016 in designated Emission Control Areas (ECA). Outside the ECA, the Tier II requirement is still performed. Compared with the requirements of Tier II and Tier I, NO<sub>x</sub> emissions must decrease by 74% to 79% and 80% respectively to meet the Tier III requirement.

In face of the new increasingly stringent emission control regulations of IMO and many coastal countries, the development and application of effective NO<sub>x</sub> emissions-reducing technology adapted in the marine diesel engine have already become the fundamental way in addition to the guidance of the government policies. This also has been becoming the most pressing

problem to be solved indispensable for the shipping industry and marine diesel engine manufacturers.

Table 1. NO<sub>x</sub> emissions limits in MARPOL73/78 Convention-Annex VI Amendment

Tier	Effective Date	NO <sub>x</sub> Limit /g·(kWh) <sup>-1</sup>		
		n<130	130≤n≤2000	n>2000
Tier I *	2000	17.0	45×n <sup>-0.2</sup>	9.84
Tier II	2011	14.36	44×n <sup>-0.23</sup>	7.66
Tier III **	2016	3.40	9×n <sup>-0.23</sup>	1.97

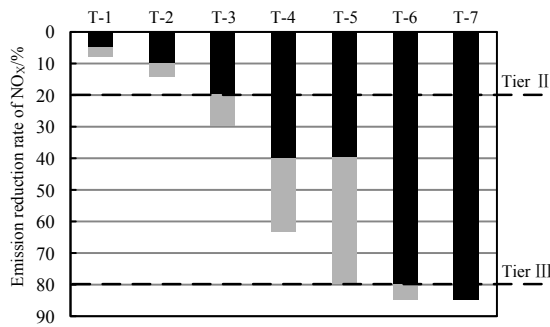
“n” refers to rated engine speed (r·min<sup>-1</sup>).

All the NO<sub>x</sub> emissions limits exclude ships with marine diesel engines less than a power output of 130 kW or ships solely for emergency use such as emergency diesel engine, lifeboat engines etc.

\*Annex VI entered into force in 2004, but it applies retroactively to new engines larger than a power output of 300kW installed on ships on or after January 1, 2000.

\*\* Tier III applies only in emission control areas.

The existing available emission reduction technologies on marine diesel engine are generally classified into two main types: internal purification and exhaust emissions after-treatment. The internal purification technologies include dual-fuel engine (Wärtsilä, 2013), alternative fuel technology, water-in-fuel emulsion and improvements of fuel and air intake system (U. S. Environmental Protection Agency, 2009). The exhaust emissions after-treatments mainly include selective catalytic reduction (SCR) method (Azzara, A., 2014), exhaust gas recirculation (EGR) (Raptotasios, S. I., 2015) and off-gas scrubbing (MAN Diesel & Turbo, 2013). The NO<sub>x</sub> reduction effects of the main available technologies are presented in Figure 1. It is acknowledged that most of these technologies cannot meet the Tier III requirement.



T-1: Supercharge and inter-cooling, T-2: After injection, T-3: Water-in-Fuel Emulsion (WFE), T-4: Humid air motors (HAM), T-5: Exhaust gas recirculation (EGR), T-6: Selective catalytic reduction (SCR), T-7: Dual-fuel (DF Gas mode)

Figure 1. NO<sub>x</sub> emission reduction effects of the main available technologies

Besides the techniques mentioned above, the homogeneous charge compression ignition (HCCI) (Maurya, R. K., 2012, Niat, P. M., 1983) technology has received increased attention for its potential to extremely reduce the emissions of NO<sub>x</sub> and PM and enhance the thermal efficiency (Xue, L., 2014, Yao, M. F., 2009). This HCCI mode is expected to be one of the effective technical solutions to meet the IMO MARPOL73/78 Convention-Annex VI Amendment Tier III requirement if it is applied to marine diesel engine. A significant feature of HCCI is that the compression ignitions are happened in multiple regions. Therefore, the temperature inside the combustion chamber is relatively uniform, the cyclic variations are small and there are no obvious high-temperature zones in combustion process. With these characteristics, the combustion roughness and fuel economy can be improved. And the formations of NO<sub>x</sub> and PM are controlled effectively. To promote the practical application of the HCCI technology, the combustion characteristic and optimal design have been studied. The cyclic variation of HCCI combustion phasing has been investigated by Maurya, R. K. (Maurya, R. K., 2013). Many experimental researches on HCCI combustion using various alternative fuels (Li, H. L., et 2007) and blended fuels (Bushby, S. R. M., 2005) have been conducted. Furthermore, both numerical simulation and experimental research show that the HCCI mode has potential for extremely low emissions of NO<sub>x</sub> and PM (Yao, M. F., 2005). However, the studies on NO<sub>x</sub> emissions reduction mechanism and influence factor for HCCI combustion are insufficient. So far, to the authors' knowledge, the study on the applicability of HCCI marine diesel engine to Tier III requirement is also a rare occurrence.

The elementary chemical reactions of diesel combustion in the cylinder are very complex and extremely large in quantity because of the multiple components and uncertainty component concentration in diesel fuel (Kim, Y. M., 2000, Turns, S. R., 2000). Consequently, the other single component fuels are always selected as replacements for diesel fuel in combustion simulation. N-heptane is one of the ideal alternative fuels which have the approximate cetane number with diesel fuel (Zheng, Z. L., 2006).

Applying the NO<sub>x</sub> chemical reaction mechanism inside the cylinder, a numerical study on the NO<sub>x</sub> formation of n-heptane for HCCI combustion process is performed taking the MAN-B&W6L23/30-A marine diesel engine as application object. On this basis, the characteristic features of NO<sub>x</sub> formation and emission are displayed. Then the effects of emission reduction under various operating conditions are analyzed and discussed.

## 2. FORMATION MECHANISM OF NO<sub>x</sub> AND COMPUTATIONAL MODEL

Based on mechanism of NO<sub>x</sub> formation and emission, NO<sub>x</sub> in diesel engine exhaust are created and discharged as a by-product of in-cylinder combustion process. According to the source of nitrogen element in NO<sub>x</sub>, NO<sub>x</sub> emissions are generally divided into three main types: thermal NO<sub>x</sub>, fuel NO<sub>x</sub> and prompt NO<sub>x</sub>. The diesel fuel usually contains trace amounts of nitrogen element. During the running of diesel engine, the fuel NO<sub>x</sub> is mainly generated by the volatile nitrogen which is formed from nitrogen element in fuel under high temperature, and the amount of volatile nitrogen is little. That is, NO<sub>x</sub> from fuel oil is so little that it can be ignored in calculation. Therefore, it is generally considered that NO<sub>x</sub> in diesel engine exhaust is almost entirely the thermal NO<sub>x</sub> and prompt NO<sub>x</sub> emissions which are produced from in-cylinder combustion chemical reaction involving the nitrogen (N<sub>2</sub>).

### 2.1 FORMATION MECHANISM OF NO<sub>x</sub>

The forms of thermal NO<sub>x</sub> include nitric oxide (NO), nitrous oxide (NO<sub>2</sub>), nitrogen dioxide (N<sub>2</sub>O), dinitrogen trioxide (N<sub>2</sub>O<sub>3</sub>), nitrogen pentoxide (N<sub>2</sub>O<sub>5</sub>), nitrogen tetroxide (N<sub>2</sub>O<sub>4</sub>) and so on, which are the major contributors to local and regional air quality issues such as acidic nitrate deposition and health impacts such as lung irritation. Among them, NO<sub>2</sub> and NO are the majority components in NO<sub>x</sub> emissions. The detailed formation mechanism of NO<sub>x</sub> is very complicated. It is generally recognized that NO<sub>x</sub> are produced in the zones of flame front and burned region. The NO<sub>x</sub> formation chain reactions are proposed by Zeldovich, Y. B. in 1946 (Palmer, W. 2000, Zeldovich, Y. B., 1947).

Pyrolysis of O<sub>2</sub> under high temperature:



Production of NO:



Production of NO<sub>2</sub>:



Another formation reaction of NO occurs in the conditions of over-rich mixture.



Under normal operating conditions of marine diesel engine, the mixture concentration is relatively low that the chemical reaction (5) is generally ignored.

For above chemical reactions, temperature is an important factor that affects the rate of chemical reaction (2), which ultimately influences the process and rates of the  $\text{NO}_x$  formation reactions. Therefore, the maximum flame temperature and maximum local flame temperature are the main influence factors that affect the formation rate of  $\text{NO}_x$  in combustion chamber. In addition, oxygen ( $\text{O}_2$ ) concentration and chemical reaction time of oxygen ( $\text{O}_2$ ) and nitrogen ( $\text{N}_2$ ) in high temperature zone also have important effects on the generation of  $\text{NO}_x$ . It is summarized that the factors which affect the chemical reaction rates of  $\text{NO}_x$  formation mainly include combustion temperature, oxygen ( $\text{O}_2$ ) concentration and chemical reaction time of oxygen ( $\text{O}_2$ ), and nitrogen ( $\text{N}_2$ ) in high temperature zone (Zhang, T., 2016).

## 2.2 THERMODYNAMIC MODEL

In this modeling, it is assumed that the thermodynamic pressure, thermodynamic temperature and composition concentration of inflammable mixture are completely uniform in combustion chamber (Figure 2). The thermodynamic state of inflammable mixture (working fluid) is determined by the pressure  $P$ , thermodynamic temperature  $T$  and mass  $M$ . And the engine working process can be simulated on the basis of energy conservation equation, mass conservation equation and state equation of ideal gas.

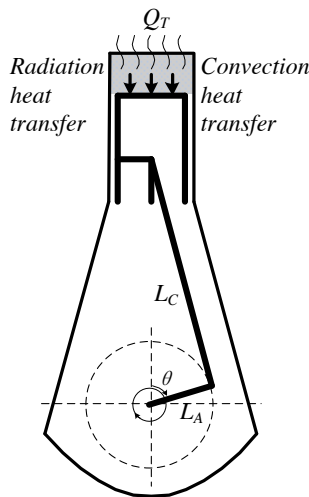


Figure 2. Modeling of HCCI diesel engines;  $L_C$ : length of connecting rod,  $L_A$ : radius of crank arm,  $\theta$ : crank angle

## 2.3 HEAT TRANSFER MODEL

The convection heat transfer and radiation heat transfer of cylinder wall are considered. The heat transfer model is described by the following empirical formula (Woschni, G., 1967):

$$Q_T = 129.8B^{-0.2}P^{0.8}T^{-0.55}w^{0.8}A_s \cdot \Delta T_m \quad (6)$$

Where  $Q_T$  is heat flux of the cylinder,  $B$  is cylinder bore,  $P$  is in-cylinder pressure,  $T$  is combustor temperature,  $A_s$  is equivalent heat transfer area,  $\Delta T_m$  is temperature difference, and  $w$  is the average velocity of in-cylinder flow.

## 2.4 PISTON MOTION

The inflammable mixture in cylinder, which is from system boundary, and residual exhaust gas are supposed to be thoroughly mixed instantaneously during the intake stroke. The change of cylinder volume caused by piston motion is simplified to a slider-crank mechanism, which is expressed as:

$$\frac{V(t)}{V_C} = 1 + \frac{\varepsilon - 1}{2} \left[ \frac{L_C}{L_A} + 1 - \cos \theta - \sqrt{\left( \frac{L_C}{L_A} \right)^2 - \sin^2 \theta} \right] \quad (7)$$

Where  $V(t)$  is cylinder volume,  $V_C$  is cylinder clearance volume,  $\varepsilon$  is geometry compression ratio

The diesel engine power is expressed as the arithmetic product of the output power of piston and transmission mechanical efficiency of connecting rod and crankshaft. The chemical reaction kinetics of  $\text{NO}_x$  formation is implemented based on CHEMKIN. The  $\text{NO}_x$  emissions, reduction rate and the other calculations are mainly performed in MATLAB.

## 3. RESULTS AND ANALYSIS

In this research, MAN-B&W6L23/30-A marine medium-speed diesel engine in marine power plant laboratory of Dalian Maritime University is chosen as the simulation object. Table 2 shows the structural specifications of MAN-B&W6L23/30-A diesel Engine.

Table 2. Structural specifications of MAN-B&W6L23/30-A diesel Engine

Items	Specification parameter
Cylinder bore/mm	225
Length of stroke/mm	300
Number of cylinders	6
Geometry compression ratio	13:1
Connecting rod length/mm	600
Cylinder clearance volume / $\text{m}^3$	$9.94 \times 10^{-4}$
Release valve open phase degree / $^\circ\text{CA}$ (BDC)	-37
Rated speed / $\text{r} \cdot \text{min}^{-1}$	900

In the simulation calculation, the TDC (Top Dead Center) of the intake stroke is set as the zero point of crank angle ( $0^\circ\text{CA}$ ). The computational domain starts from the starting point of compression stroke, ends with terminal

point of expansion stroke. So the range of crank angle is from 180°CA to 540°CA. The fuel injection way adopts the port fuel injection.

### 3.1 IMPACT ANALYSIS OF COMPRESSION RATIO

Figure 3 and Figure 4 show the  $\text{NO}_x$  concentration in cylinder and  $\text{NO}_x$  emissions with the compression ratios. According to the observation, the compression ratio affects the  $\text{NO}_x$  emissions greatly. With the rise of compression ratio, the  $\text{NO}_x$  emissions increase continuously and the increase rate is accelerated. When the compression ratio is less than 12, the  $\text{NO}_x$  emissions tend towards zero. When the compression ratio exceeds 18, the  $\text{NO}_x$  emissions have already reached  $4.5\text{g}\cdot(\text{kWh})^{-1}$  that achieves the level of the conventional diesel engine. As shown in Figure 5 and Figure 6, this phenomenon is result from the case that the high compression ratio would cause higher combustor temperature and  $\text{O}_2$  concentration in cylinder. The percentage of NO in the  $\text{NO}_x$  emissions rises from 69.4% at compression ratio of 11 to 99.4% at compression ratio of 18.

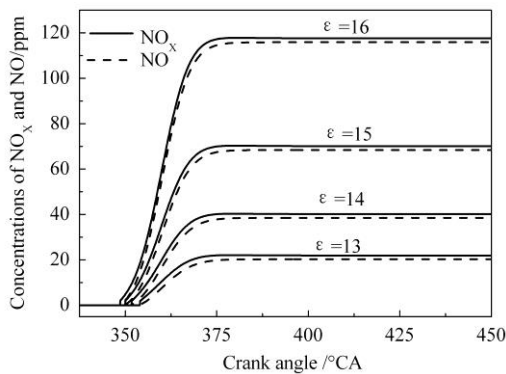


Figure 3. Effect of the compression ratio on concentration of  $\text{NO}_x$  and NO in cylinder. (Intake pressure  $p_0=0.25$  MPa, intake air temperature  $T_0=320$  K, excess air coefficient  $\lambda=2.8$ , engine speed  $n=900$  r·min $^{-1}$ .)

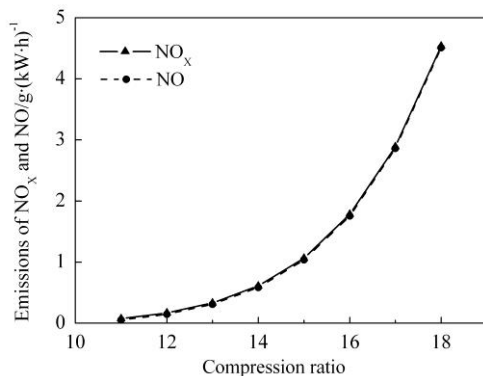


Figure 4. Effect of the compression ratio on  $\text{NO}_x$  emissions. (Intake pressure  $p_0=0.25$  MPa, intake air temperature  $T_0=320$  K, excess air coefficient  $\lambda=2.8$ , engine speed  $n=900$  r·min $^{-1}$ .)

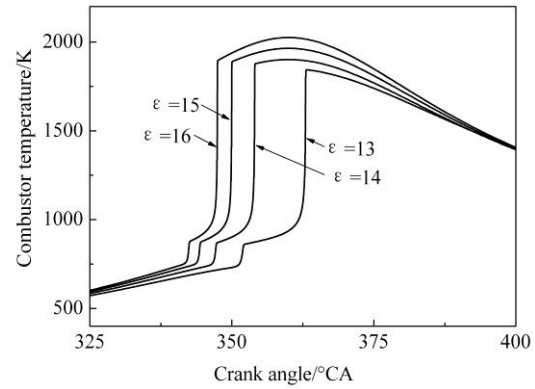


Figure 5. Effect of the compression ratio on combustor temperature. (Intake pressure  $p_0=0.25$  MPa, intake air temperature  $T_0=320$  K, excess air coefficient  $\lambda=2.8$ , engine speed  $n=900$  r·min $^{-1}$ .)

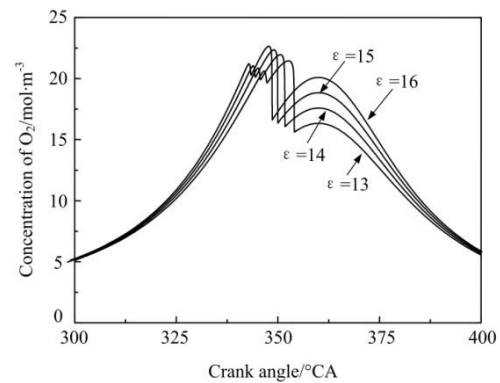


Figure 6. Effect of the compression ratio on concentration of  $\text{O}_2$  in cylinder. (Intake pressure  $p_0=0.25$  MPa, intake air temperature  $T_0=320$  K, excess air coefficient  $\lambda=2.8$ , engine speed  $n=900$  r·min $^{-1}$ .)

The modern marine diesel engine commonly adopts high compression ratio which is unfavorable for the reduction of  $\text{NO}_x$  emissions in HCCI mode. But due to high combustion efficiency of this combustion mode, the  $\text{NO}_x$  emissions have the potential to be reduced moderately in the operating condition of low compression ratio without affecting the thermal efficiency and fuel economical efficiency significantly.

### 3.2 IMPACT ANALYSIS OF INTAKE AIR TEMPERATURE

Figure 7 and Figure 8 show the  $\text{NO}_x$  concentration in the cylinder and  $\text{NO}_x$  emissions with the intake air temperature. The  $\text{NO}_x$  emissions increase with the rise of intake air temperature in the case of unchangeableness of the other initial conditions. The effects of the intake air temperature on combustor temperature and concentration of  $\text{O}_2$  are given in Figure 9 and Figure 10, higher intake air temperature often result in higher combustor temperature and lower concentration of  $\text{O}_2$ . It is obvious

that the combustor temperature plays more important role in the chemical reaction rate of  $\text{NO}_x$  formation under the simulating operating conditions.

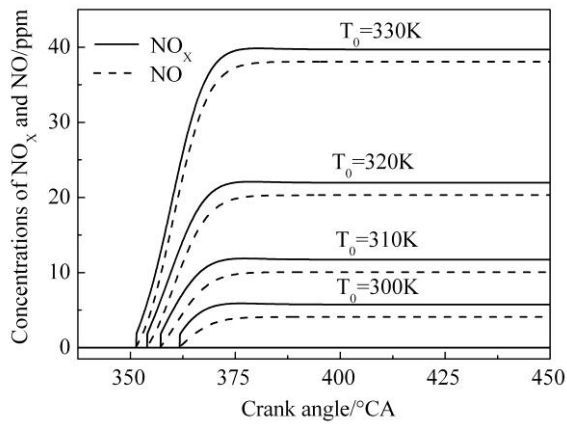


Figure 7. Effect of the intake air temperature on concentration of  $\text{NO}_x$  and NO in cylinder. (Intake pressure  $p_0=0.25$  MPa, compression ratio  $\varepsilon=13$ , excess air coefficient  $\lambda=2.8$ , engine speed  $n=900$  r·min<sup>-1</sup>.)

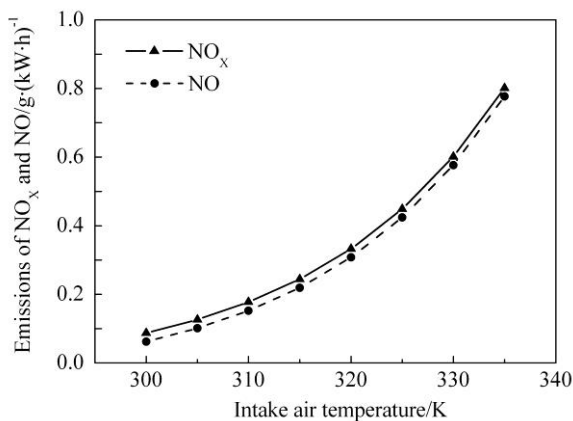


Figure 8. Effect of the intake air temperature on  $\text{NO}_x$  emissions. (Intake pressure  $p_0=0.25$  MPa, compression ratio  $\varepsilon=13$ , excess air coefficient  $\lambda=2.8$ , engine speed  $n=900$  r·min<sup>-1</sup>.)

In the intake air temperature range of 300K to 335K, the  $\text{NO}_x$  emissions are consistently lower than  $0.8 \text{ g} \cdot (\text{kW} \cdot \text{h})^{-1}$ . The most mass of chemical component is NO which contributes the  $\text{NO}_x$  emissions of percentage of about 71.2% to 97.0%. The other components in  $\text{NO}_x$  emissions are mainly  $\text{NO}_2$ . Moreover, the percentage of the NO increases with the intake air temperature. That is because the temperature has obvious influence on the NO formation reaction.

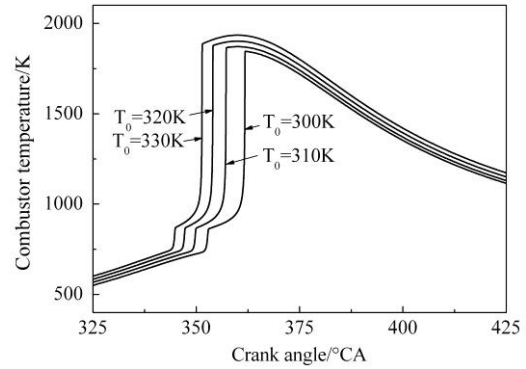


Figure 9. Effect of the intake air temperature on combustor temperature. (Intake pressure  $p_0=0.25$  MPa, compression ratio  $\varepsilon=13$ , excess air coefficient  $\lambda=2.8$ , engine speed  $n=900$  r·min<sup>-1</sup>.)

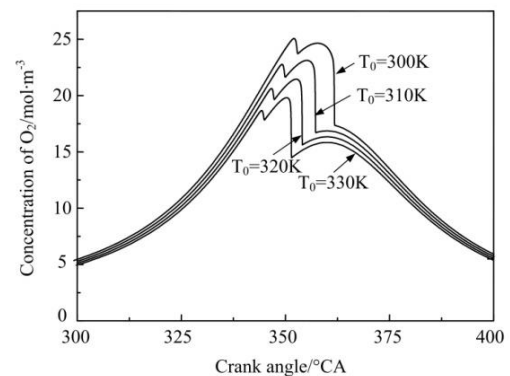


Figure 10. Effect of the intake air temperature on concentration of  $\text{O}_2$  in cylinder. (Intake pressure  $p_0=0.25$  MPa, compression ratio  $\varepsilon=13$ , excess air coefficient  $\lambda=2.8$ , engine speed  $n=900$  r·min<sup>-1</sup>.)

### 3.3 IMPACT ANALYSIS OF INTAKE PRESSURE

Figure 11 and Figure 12 show the  $\text{NO}_x$  concentration in the cylinder and  $\text{NO}_x$  emissions with the intake pressure. It is shown that the  $\text{NO}_x$  emissions exhibit an approximate linear increase with the intake pressure. When the intake pressure is in the range of 0.15 MPa to 0.45 MPa, the  $\text{NO}_x$  emissions are relatively low. It is also shown that NO is the major chemical component in  $\text{NO}_x$  emissions. The percentage of NO decreases from 95.7% to 86.9% in the chosen range of intake pressure.

Figure 13 and Figure 14 show the effects of the intake air temperature on combustor temperature and concentration of  $\text{O}_2$  in cylinder. The increase of intake pressure leads to rising of concentration of  $\text{O}_2$  and extension of chemical reaction time of  $\text{O}_2$  and  $\text{N}_2$  in high temperature zone. Due to the combined effects of the two factors, the  $\text{NO}_x$  emissions tend to grow rapidly.

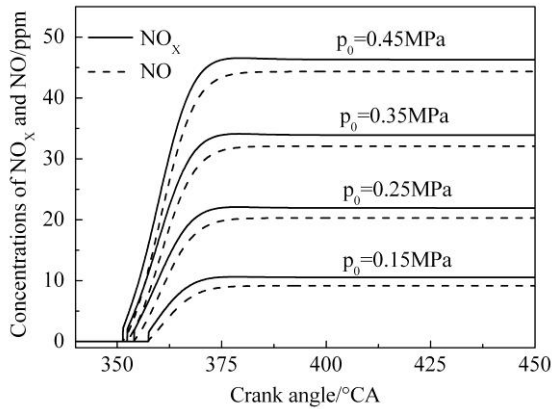


Figure 11. Effect of the initial pressure on concentration of  $\text{NO}_x$  and  $\text{NO}$  in cylinder. (Intake air temperature  $T_0=320$  K, compression ratio  $\varepsilon=13$ , excess air coefficient  $\lambda=2.8$ , engine speed  $n=900$  r·min $^{-1}$ .)

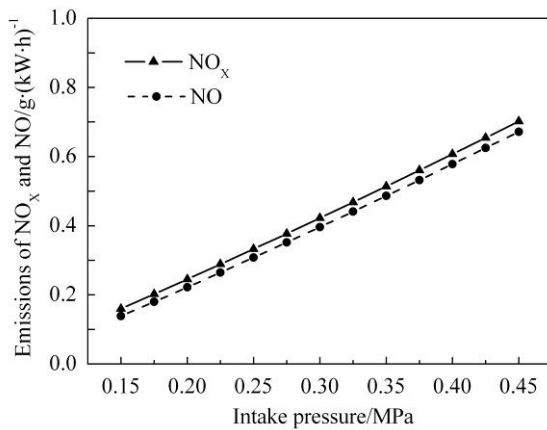


Figure 12. Effect of the initial pressure on  $\text{NO}_x$  emissions. (Intake air temperature  $T_0=320$  K, compression ratio  $\varepsilon=13$ , excess air coefficient  $\lambda=2.8$ , engine speed  $n=900$  r·min $^{-1}$ .)

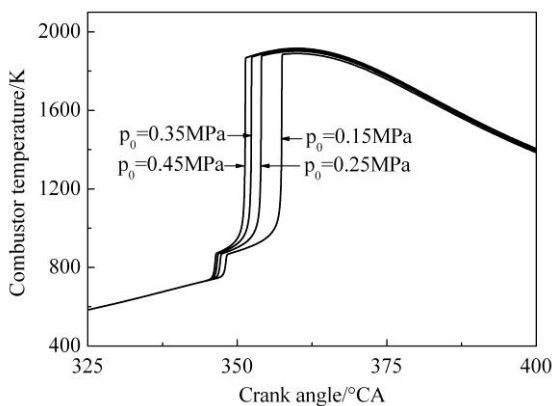


Figure 13. Effect of the initial pressure on combustor temperature. (Intake air temperature  $T_0=320$  K, compression ratio  $\varepsilon=13$ , excess air coefficient  $\lambda=2.8$ , engine speed  $n=900$  r·min $^{-1}$ .)

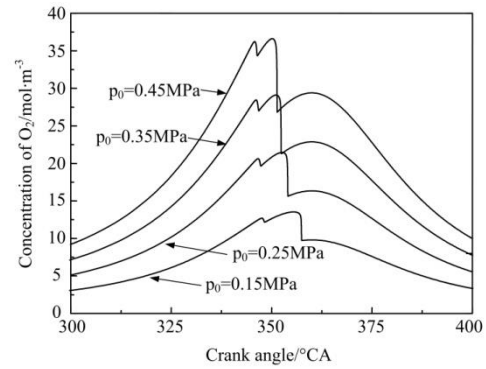


Figure 14. Effect of the initial pressure on concentration of  $\text{O}_2$  in cylinder. (Intake air temperature  $T_0=320$  K, compression ratio  $\varepsilon=13$ , excess air coefficient  $\lambda=2.8$ , engine speed  $n=900$  r·min $^{-1}$ .)

### 3.4 IMPACT ANALYSIS OF EXCESS AIR COEFFICIENT

Figure 15 and Figure 16 show the  $\text{NO}_x$  concentration in the cylinder and  $\text{NO}_x$  emissions with the excess air coefficient. The impacts of excess air ratio to the  $\text{NO}_x$  emissions are remarkably. Overall, the  $\text{NO}_x$  emissions decrease with the increase of the excess air ratio. When the excess air ratio is less than 2.3, the  $\text{NO}_x$  emissions decrease very quickly but far exceed the emission level of conventional diesel engine, and the components in  $\text{NO}_x$  emissions are almost entirely  $\text{NO}$ . When the excess air ratio is greater than 2.3, the  $\text{NO}_x$  emissions decrease slowly relatively. The other components of  $\text{NO}_x$  emissions, except  $\text{NO}$ , account for a certain proportion. When the excess air ratio is greater than 3.2, the  $\text{NO}_x$  emissions have approached zero.

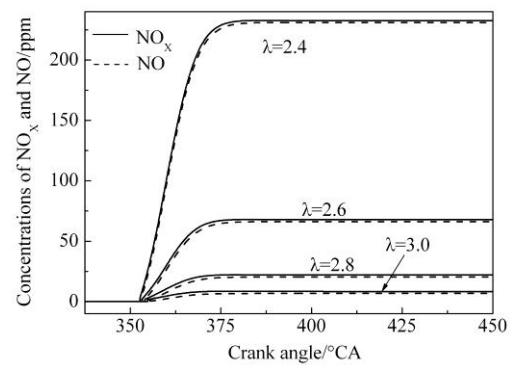


Figure 15. Effect of the excess air coefficient on concentration of  $\text{NO}_x$  and  $\text{NO}$  in cylinder. (Intake air temperature  $T_0=320$  K, intake pressure  $p_0=0.25$  MPa, compression ratio  $\varepsilon=13$ , engine speed  $n=900$  r·min $^{-1}$ .)

Figure 17 and Figure 18 show the effects of the intake air temperature on combustor temperature and concentration of  $\text{O}_2$  in cylinder. The simulation results confirm that, with the increase of the excess air ratio, the declines of combustor temperature and chemical reaction time of  $\text{O}_2$  and  $\text{N}_2$  in high temperature zone are significant. But the increase of concentration of  $\text{O}_2$  is generally less

pronounced. Their synthesis results are that the  $\text{NO}_x$  emissions are cut down dramatically at a larger excess air coefficient. Therefore, it can be seen that HCCI engine has extraordinarily lower  $\text{NO}_x$  emissions with the dilute homogeneous air and fuel mixture. The  $\text{NO}_x$  emissions of HCCI marine diesel engine can be controlled by adjusting the excess air ratio.

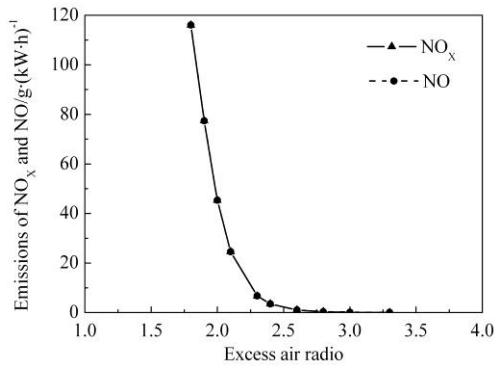


Figure 16. Effect of the excess air coefficient on  $\text{NO}_x$  emissions. (Intake air temperature  $T_0=320$  K, intake pressure  $p_0=0.25$  MPa, compression ratio  $\varepsilon=13$ , engine speed  $n=900$  r·min $^{-1}$ .)

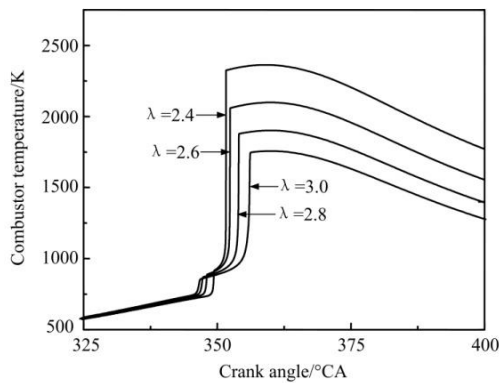


Figure 17. Effect of the excess air ratio on combustor temperature. (Intake air temperature  $T_0=320$  K, intake pressure  $p_0=0.25$  MPa, compression ratio  $\varepsilon=13$ , engine speed  $n=900$  r·min $^{-1}$ .)

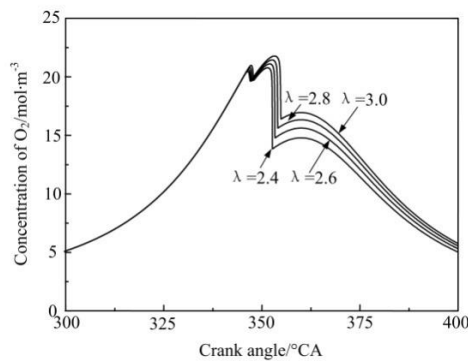


Figure 18. Effect of the excess air ratio on concentration of  $\text{O}_2$  in cylinder. (Intake air temperature  $T_0=320$  K, intake pressure  $p_0=0.25$  MPa, compression ratio  $\varepsilon=13$ , engine speed  $n=900$  r·min $^{-1}$ .)

### 3.5 IMPACT ANALYSIS OF ENGINE SPEED

Figure 19 and Figure 20 show the  $\text{NO}_x$  concentration in the cylinder and  $\text{NO}_x$  emissions with the engine speed. It is shown that the  $\text{NO}_x$  emissions decrease continually with the increase of engine speed, but all in low level in the simulated conditions. The percentages of NO in  $\text{NO}_x$  emissions are in the range of 97.3% to 91.6%.

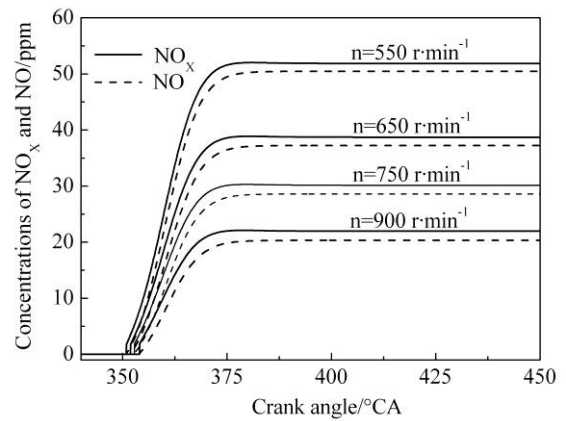


Figure 19. Effect of the engine speed on concentration of  $\text{NO}_x$  and NO in cylinder. (Intake air temperature  $T_0=320$  K, intake pressure  $p_0=0.25$  MPa, excess air coefficient  $\lambda=2.8$ , compression ratio  $\varepsilon=13$ .)

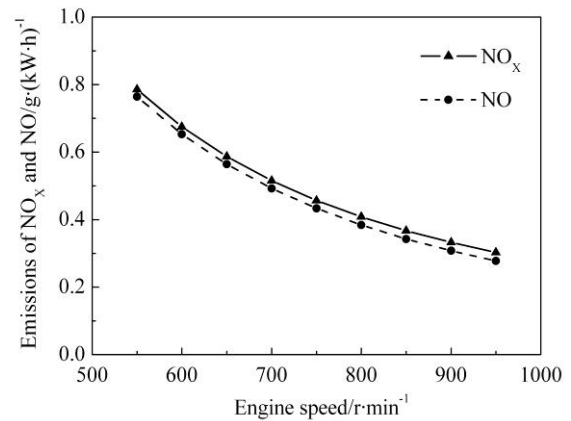


Figure 20. Effect of the engine speed on  $\text{NO}_x$  emissions. (Intake air temperature  $T_0=320$  K, intake pressure  $p_0=0.25$  MPa, excess air coefficient  $\lambda=2.8$ , compression ratio  $\varepsilon=13$ .)

Figure 21 and Figure 22 display the effects of the engine speed on combustor temperature and concentration of  $\text{O}_2$  in cylinder. The rise of engine speed shortens the chemical reaction time of  $\text{O}_2$  and  $\text{N}_2$  in high temperature zone. But the combustor temperature and concentration of  $\text{O}_2$  are not changed obviously in the crank angle of exceeding  $350^\circ\text{CA}$ . The  $\text{NO}_x$  emissions and the percentage of the NO emissions decrease as a whole.

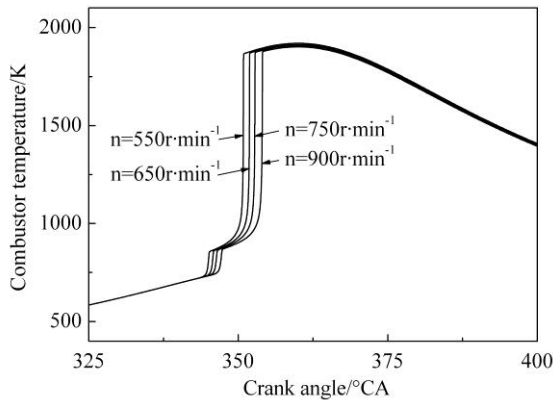


Figure 21. Effect of the engine speed on combustor temperature. (Intake air temperature  $T_0=320$  K, intake pressure  $p_0=0.25$ MPa, excess air coefficient  $\lambda=2.8$ , compression ratio  $\varepsilon=13$ .)

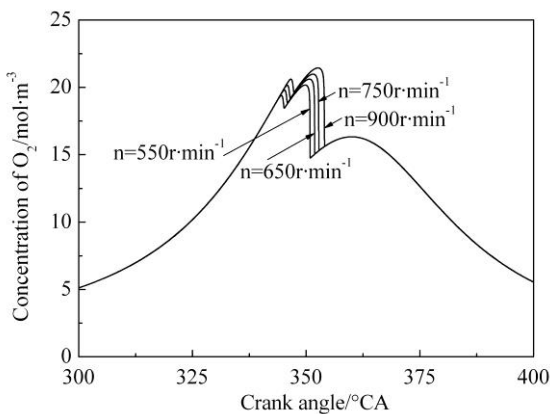


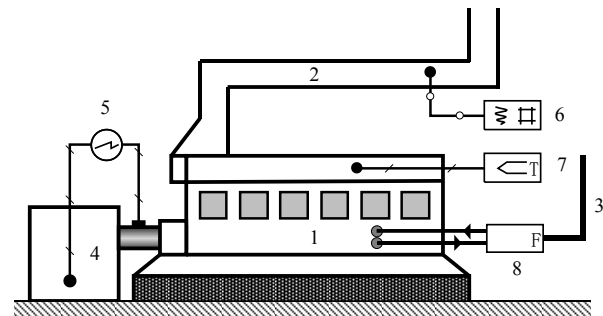
Figure 22. Effect of the engine speed on concentration of  $O_2$  in cylinder. (Intake air temperature  $T_0=320$  K, intake pressure  $p_0=0.25$ MPa, excess air coefficient  $\lambda=2.8$ , compression ratio  $\varepsilon=13$ .)

#### 4. ANALYSIS OF $NO_x$ EMISSIONS REDUCTIONS

At present, the conventional compression-ignition engine diesel and spark-ignition engine, each of which has reached its limit in improvement of fuel efficiency and decrease of harmful emission. The HCCI technology with the features of homogeneous mixture and low temperature reaction are expected to not only improve the engine cycle thermal efficiency, but significantly reduce the pollutant emission of  $NO_x$ .

In this paper, the MAN-B&W6L23/30-A marine diesel engine is chosen as application object. The experimental data of  $NO_x$  emissions are obtained from Dalian Maritime University. The schematic diagram of the diesel engine test is shown in Figure 23 (Xu, L. P., 2003). The effects of  $NO_x$  emissions reduction on n-heptane HCCI mode which is applied to the marine diesel engine

are analyzed with the same structural specifications and operating conditions. As shown in Table 3 and Table 4, the  $E_1$  and  $E_2$  cycle of marine diesel engine are the test cycle modes according to the IMO MARPOL73/78 Convention-Annex VI.



1: MAN-B&W6L23/30-A marine diesel engine  
2: exhaust pipe  
3: fuel pipe  
4: hydraulic dynamometer system  
5: power measuring instrument  
6: flue gas analyzer CLD/PMD  
7: temperature tester  
8: fuel consumption instrument.  
Figure 23. Schematic diagram of the diesel engine test

In Table 3 and Table 4, it is shown that the effect of  $NO_x$  emissions reduction on HCCI mode applied to the marine diesel engine is extraordinary obvious. Under these operating conditions, the average  $NO_x$  reduction rate can surpass 95% and the emission values can fully meet the limits of Tier III requirement.

#### 5. CONCLUSIONS

- (1) Results from simulation show that  $NO$  usually accounts for the largest proportion in  $NO_x$  formation and emission of HCCI mode marine diesel engine under various simulated conditions. The combustor temperature always plays more important role in the proportion of  $NO$ .
- (2) The operating conditions of the low compression ratio, low intake air temperature, low inlet pressure, large excess air coefficient and high engine speed are beneficial in reducing  $NO_x$  formation and emission of HCCI diesel engine. The above changes of operating conditions primarily influence the factors of combustion temperature, oxygen concentration and chemical reaction time of oxygen and nitrogen in high temperature zone, which affect the chemical reaction rate of  $NO_x$  generation.
- (3) The HCCI combustion mode tends to have extremely remarkable low  $NO_x$  emissions performance. Compared with the experimental data of actual marine diesel engine, simulation results show that the emission reduction rate of  $NO_x$  can achieve an average of more than 95% in the n-heptane HCCI mode.



Table 3. Effect of NO<sub>x</sub> emissions reduction of marine diesel engine in E<sub>2</sub> cycle on HCCI mode

Items	Operating condition 1	Operating condition 2	Operating condition 3	Operating condition 4
Engine speed /r·min <sup>-1</sup>	900	900	900	900
Scavenging temperature/°C	41	50	50	56
Scavenging pressure/bar	0.03	0.36	1.35	1.8
Experimental data of NO <sub>x</sub> emissions / g·(kWh) <sup>-1</sup>	10.58	9.72	9.72	9.13
HCC simulation data of NO <sub>x</sub> emissions/ g·(kWh) <sup>-1</sup>	0.053	0.194	0.399	0.566
NO <sub>x</sub> emissions limits of Tier III requirement / g·(kWh) <sup>-1</sup>	1.883	1.883	1.883	1.883
NO <sub>x</sub> reduction rate /%	99.5	98.0	95.9	93.8

Table 4. Effect of NO<sub>x</sub> emissions reduction of marine diesel engine in E<sub>3</sub> cycle on HCCI mode

Items	Operating condition 1	Operating condition 2	Operating condition 3	Operating condition 4
Engine speed /r·min <sup>-1</sup>	565	713	820	900
Scavenging temperature/°C	47	48	51	56
Scavenging pressure/bar	0.25	0.71	1.38	1.8
Experimental data of NO <sub>x</sub> emissions / g·(kWh) <sup>-1</sup>	11.89	10.77	10.13	9.13
HCC simulation data of NO <sub>x</sub> emissions/ g·(kWh) <sup>-1</sup>	0.333	0.377	0.476	0.566
NO <sub>x</sub> emissions limits of Tier III requirement / g·(kWh) <sup>-1</sup>	2.095	1.986	1.923	1.883
NO <sub>x</sub> reduction rate /%	97.2	96.5	95.3	93.8

## 6. ACKNOWLEDGEMENTS

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